THE PROPERTIES OF MATTER

Warm-Up a/ Is crumbling a piece of paper an example of a physical change or a chemical change b/ How do you know.

DO NOT touch the 7th grade projects... Nelson

PHYSICAL AND CHEMICAL PROPERTIES

A. PROPERTIES ARE USED TO IDENTIFY SUBSTANCES.

B. TWO KINDS OF PROPERTIES: 1. PHYSICAL PROPERTIES 2. CHEMICAL PROPERTIES

C. PHYSICAL PROPERTIES

- **1. CAN BE OBSERVED AND MEASURED.**
- 2. CHANGING PHYSICAL PROPERTY DOESN'T CHANGE THESUBSTANCE.
- **3. USED TO IDENTIFY SUBSTANCES WITHOUT DESTROYING THEM.**

EXAMPLES OF PHYSICAL PROPERTIES INCLUDE: a/ STATE OF MATTER (s, l, g) **b/ COLOR** c/ DENSITY d/ SHAPE AND SIZE ...MANY MORE!

D. <u>PHYSICAL CHANGE</u>: A CHANGE IN THE SIZE, SHAPE, COLOR, STATE OF MATTER (s,l,g),OF A SUBSTANCE.

E. MIXTURES CAN BE SEPERATED BY PHYSICAL MEANS:

1. STRAINING / FILTERING (SIZE)



2. DISSOLVING



3. EVAPORATION



Water evaporates slowly as it turns into a gas and mixes with the air.



Water evaporates quickly when it is heated. As the water boils, it turns into steam.

4. MAGNETISM



5. CENTRIFUGE (SPINNING)



6. FLOATING & SINKING (DENSITY)



F. CHEMICAL PROPERTIES

CHEMICAL CHANGE: A/ OLD SUBSTANCES ARE LOST AND NEW SUBSTANCES ARE GAINED.

B/ ELEMENTS/ATOMS ARE REARRANGED.

C/ THERE IS EVIDENCE OF A CHEMICAL REACTION.

BURNING EXPLOSION FIRE GAS GIVEN OFF

2. CHEMICAL PROPERTIES: **DESCRIBE HOWA SUBSTANCE WILL REACT AND CHANGE IN THE PRESENCE OF OTHER SUBSTANCES.**

3. EXAMPLES OF CHEMICAL PROPERTIES A/ COMBUSTIBLE-EXPLOSIVE B/ FLAMMABLE-FIRE C/ CORROSIVE-EATS THRU METAL (ACIDS) D/ CAUSTIC-EATS THRU FLESH (BASES)

E/ RUST/TARNISH-OXYGEN REACTS WITH METALS F/ REACTIVE G/NON-REACTIVE-INERT H/TOXIC-POISONOUS

CHEMICAL EQUATIONS ARE USED TO DESCRIBE CHEMICAL CHANGES:

$Mg + O_{2} \rightarrow 2Mg O$ REACTANTS \rightarrow PRODUCTS OLD SUBSTANCE NEW SUBSTANCE!

NEW PROPERTIES!